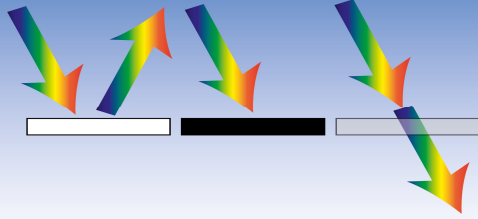


Lecture 2 – How does Light Interact with the Environment?



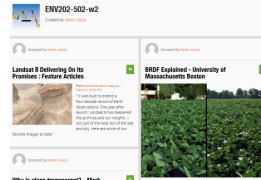
Dr Karen Joyce
School of Environmental and Life Sciences
Bldg Purple 12.3.09

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1

Treasure Hunt

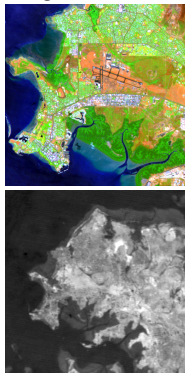
- Find and scan all 11 QR codes
- Choose one to watch / read in detail
- Post the key points as a 'reaction' to <http://www.scoop.it/t/env202-502-w2> (need to sign in first)



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2

EMR and Remote Sensing



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3

EMR Principles and Properties

- Understanding how EMR interacts with the feature you are interested in is the basis for interpreting remotely sensed data
- This also enables you to select appropriate remotely sensed data
- Leads to visual & quantitative interpretation of images and other data sets to IDENTIFY features and MEASURE biophysical properties of interest

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4

EMR Wave and Particle Theories

- Any material with molecular motion ($T > 0K$) emits radiant energy
- Can earth surface features be differentiated based on the amount of EMR they emit?
- EMR radiates according to wave theory

$$c = v\lambda$$

As c remains constant, if v increases, λ must decrease (and vice versa)

where

c = speed of light ($3 \times 10^8 \text{m} \cdot \text{sec}^{-1}$)

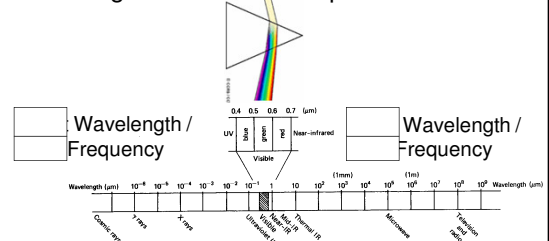
λ = wavelength of EMR ($\mu\text{m} = \text{micro-m} = 10^{-6}\text{m}$)

v = frequency of EMR (sec^{-1})

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5

Regions of the EMR Spectrum



- Wavelengths and the spectrum - EMR units
- Frequencies (instead of wavelengths as units?)
- Energy levels

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6

Particle Theory

- EMR is composed of discrete units (photons or quanta)
- Determines energy level of EMR
- Planck's equation for the EMR emitted at a specific wavelength and frequency:

$Q = h \nu$

$Q = h c / \lambda$

where

Q = radiant energy of quanta (joules)
h = Planck's constant (6.626×10^{-34} joules.sec)
ν = frequency of EMR (sec^{-1})

Remember: $c = \nu \lambda$ or: $\nu = c / \lambda$

As h & c are constants, energy is inversely proportional to wavelength

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Energy, Wavelength & Frequency


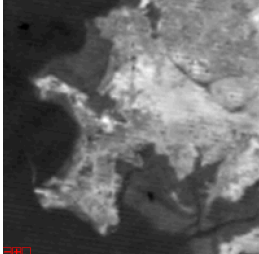
- Wavelength (inverse) and frequency (direct) relationships with EMR energy levels

(Diagram source: The Light Measurement handbook <http://www.intl-light.com/handbook/>)

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Energy & Wavelength

- Why is the Thermal Infrared image more blurry than the Mid Infrared image?

MIR
TIR

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EMR Units

- Airborne and satellite imaging systems record the amount of EMR being reflected or emitted from a target
- Amount of EMR reaching a sensor is in one of the following units:

TERM	UNIT	DESCRIPTION
radiant energy (Q)	joule	energy per quanta
radiant flux (ϕ)	watt (joule.second ⁻¹)	energy per unit time
Radiant flux density		
irradiance (E,Q)	watt.m ⁻²	incident flux per unit area over all angles
radiance (L)	watt.m ⁻² .sr ⁻¹	incident flux per unit area at specific solid/3-D angle
spectral radiance (L)	watt.m ⁻² .sr ⁻¹ .µm ⁻¹	incident flux per unit area at specific angle over a set range of wavelengths

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EMR Units

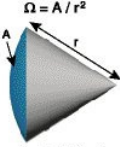
$\Omega = A / r^2$


Fig. 7.2 Solid angle




Fig. 7.1

Note: sr = Steradian, refers to a unit solid angle (a 3D angle) that relates to a camera or sensor's field of view

(Diagram source: The Light Measurement handbook <http://www.intl-light.com/handbook/>)

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Stefan Boltzman Law

- Provides a basis for distinguishing objects based on measurements of emitted energy
- Temperature controls the amount of emitted energy from an object
- The amount of energy a blackbody radiates (M) is a function of its temperature (T)

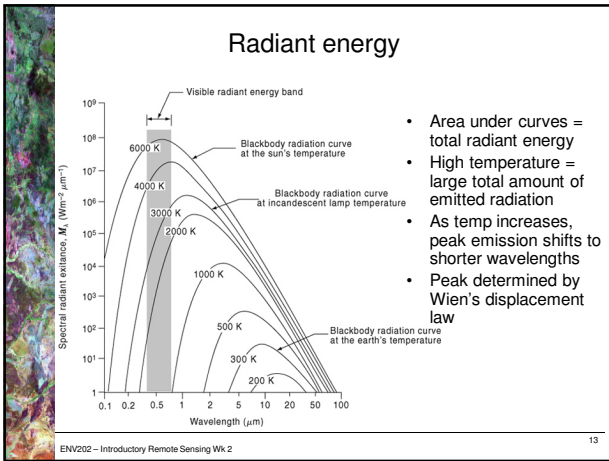
$M = \sigma T^4$

Emitted energy increases very quickly with increases in Temp

where, M = total spectral radiant exitance (watts.m⁻²)
σ = Stefan-Boltzman constant (5.6697×10^{-8} watts.m⁻² K⁻⁴)
T = temperature in K

- Note the threshold for reflection/emission is around 3µm, i.e. most EMR beyond this emitted, while EMR at < 3µm is reflected sunlight

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Wien's Displacement Law

- Use to select optimum wavelength of EMR to record for monitoring specific targets
- Wavelength of maximum emitted energy depends on an object's temperature
- The wavelength (λ) of maximum spectral radiant exitance (M) is **inversely related to its temperature (T):**

$$\lambda_{\text{max}} = A / T$$

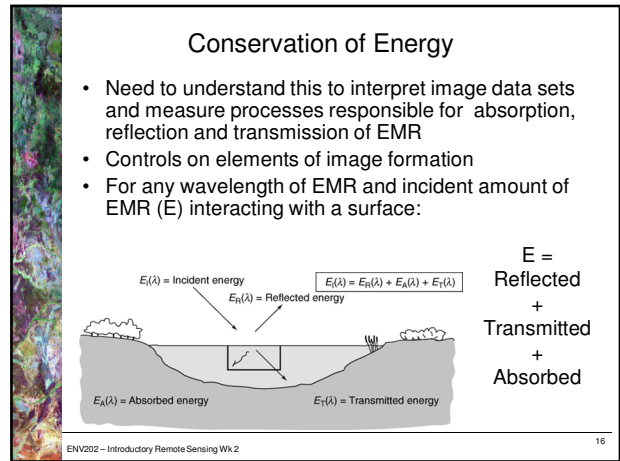
Higher temp = Shorter wavelength

where, λ_{max} = wavelength of maximum spectral radiant exitance (μm)
 $A = 2898 \mu\text{m}\cdot\text{K}$
 $T =$ temperature in K

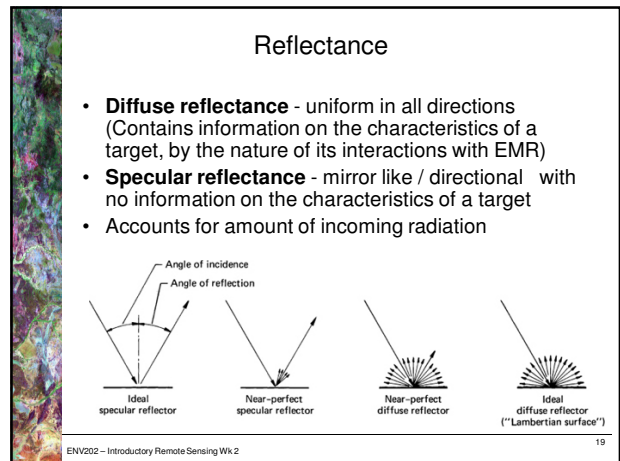
Color Scale of Temperature	
Color	Temperature
dark red	500 - 570 - 100
red	570 - 700 - 150
orange	700 - 850 - 200
yellowish red	1050 - 1150 - 220 - 1420
incandescent white heat	1250 - 1350 - 1520 - 1620
white heat	1450 - 1550 - 1720 - 1820

Sources: Process Associates of America & Handbook of Chemistry & Physics, 1924 14

- ### Recap
- As wavelengths decrease,
 - frequency increases (wave theory),
 - energy increases
 - Emitted energy increases very quickly with increases in temperature (Stefan Boltzman Law)
 - As temperatures increase, wavelength of maximum emitted energy decreases (Wien's Displacement Law)
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- ### EMR Interactions
- Reflection
 - Re-direction of light striking non-transparent surface
 - Strength of reflection – type of surface
 - Diffuse – smooth
 - Specular (Lambertian) – rough
 - Absorption
 - Energy of the photon is taken up by the feature and converted to other forms of energy (e.g. used for photosynthesis)
 - Transmission
 - Light passing through material without much attenuation
 - Water most affected by transmission of light
 - Plant leaves
-
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Reflectance and Geometry

- Reflected EMR depends on angles of incident light and camera position

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EMR Interactions in the Atmosphere

- How does the atmosphere alter the "quality" of satellite/ airborne images and aerial photographs?
- Identification of EMR interactions affecting an airborne or satellite image
- The effect of the atmosphere on the transmission of EMR to and from the earth's surface by scattering and absorption processes is a function of
 - Path length
 - EMR wavelength
 - Atmospheric conditions

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Scattering

- Rayleigh**
 - particles smaller in diameter than the EMR wavelength
 - inversely proportional to 4th power of wavelength
 - air molecules scatters short wavelengths (blue sky)
- Mie**
 - Particles equal in size to wavelength
 - Inversely proportional to 0.6-2nd power of wavelength
 - Water vapour, dust (haze), causes sky to take on reddish appearance
- Non-selective**
 - Particles have greater dimensions than wavelength
 - Scatters all wavelengths
 - Water Droplets and ice (fog and clouds), causes white appearance
- All scattering produces "additive path radiance"

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Absorption

- Reduces the amount of incident solar radiation and reflected or emitted radiation traveling to the sensor
- Loss of EMR energy to atmospheric constituents
- Major absorbers:
 - H₂O, CO₂, O₃, O₂, N₂, O, N
- Minor absorbers:
 - NO, N₂O, CO, CH₄

Source: <http://modarch.gsfc.nasa.gov/MODIS/ATM/solar.html>

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Atmospheric Windows

- Regions of the EMR spectrum where there is limited absorption of EMR by atmospheric constituents.
- EMR reflected or transmitted through the atmosphere measured by sensors in these spectral regions.

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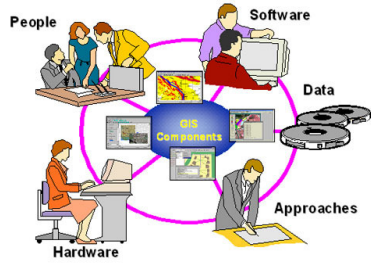
Your Job

- Annotate this image with arrows to indicate areas of reflection, absorption, transmission, scattering, and refraction

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Student Choice: GIS vs. Remote Sensing

- Go to <http://padlet.com/wall/env202-502-rsgis> and post what comes to mind



http://www.polkcitymaps.org/GIS_Components.jpg